

Polymers in Soft Matter: From Structure to Emerging Applications

Part 1: Polymer Structure and Theoretical Models

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"You Either Die A Hero, Or You Live Long Enough To See Yourself Become The Villain"



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"You Either Die A Hero, Or You Live Long Enough To See Yourself Become The Villain"



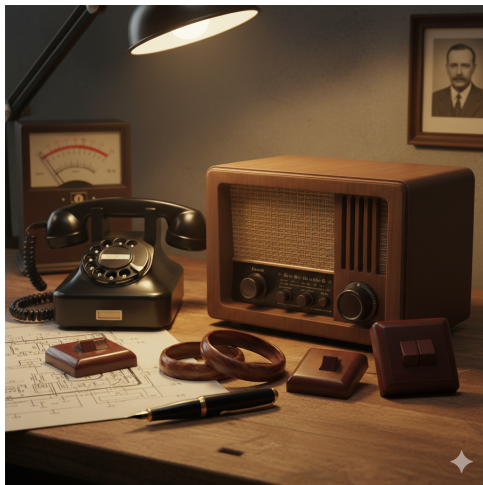
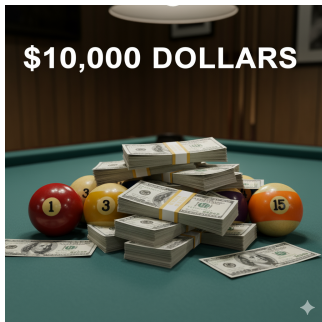
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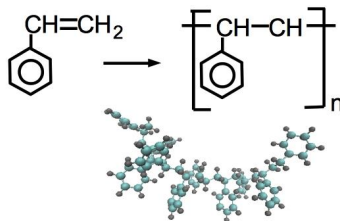
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Celluloid (1869) and Bakelite (1909)

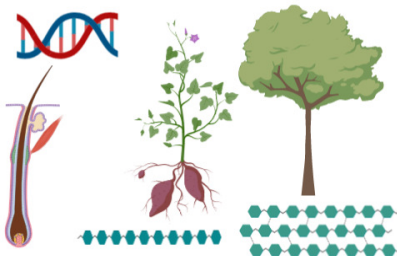


What is polymer?

- ▶ macromolecule consisting of monomers
- ▶ prepared by polymerization
- ▶ degree of polymerization n



- Synthetic
- Biopolymers



*the picture was created with BioRender



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Thermoplastics, thermosets, elastomers



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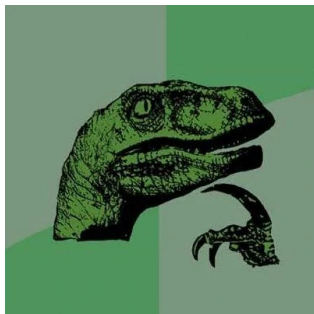


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Find one key property which commercial polymers have in common



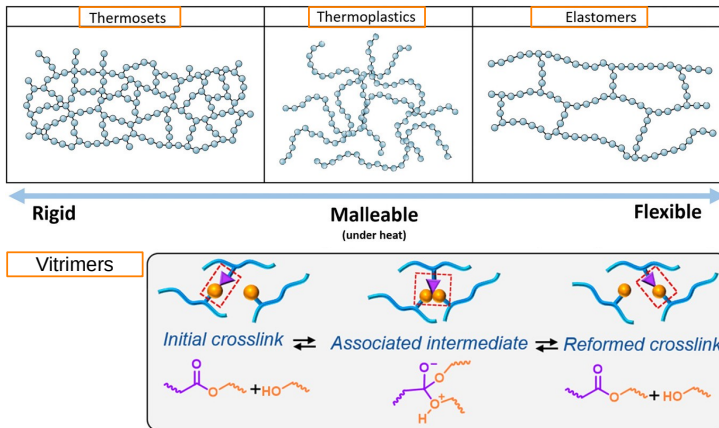
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Thermoplastics, thermosets, elastomers



[1] Cuttaz, E. A. et al.; Adv. Healthcare Mater. 2024, 13, 2304447

[2] Chen, Q.; Yang, Y.; Yu, Y.; Xu, H.; Journal of Chemical Education 2021 98 (4), 1429-1435

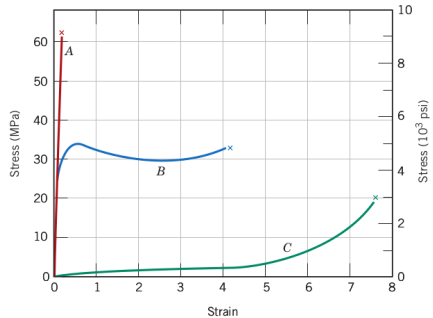
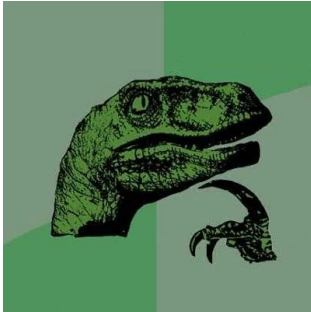


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Which stress-strain curve belongs to elastomer, thermoset and thermoplastic?

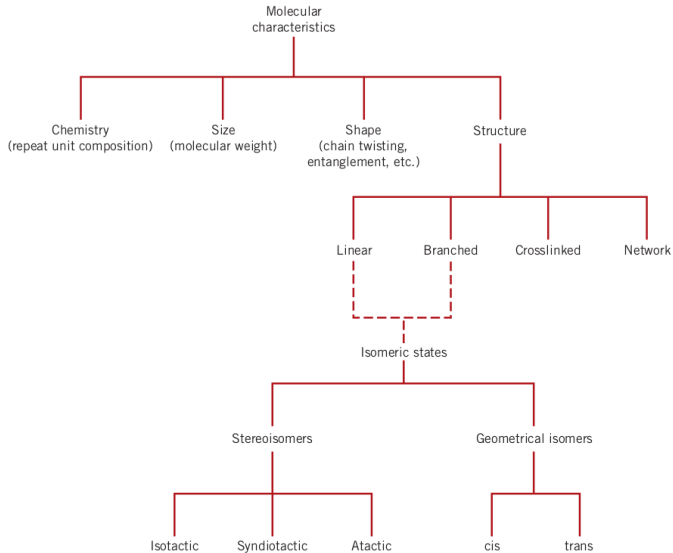


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[1] Callister, W. D., Rethwisch, D. G. (2018).
Materials Science and Engineering: An Introduction
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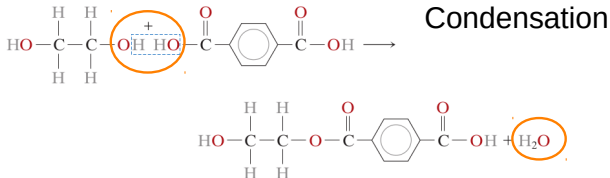
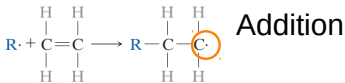
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Polymer chemistry

$\begin{array}{c} \text{H} & \text{H} \\ & \\ -(\text{C}-\text{C})_n \\ & \\ \text{H} & \text{R} \end{array}$	 1 PET	 2 HDPE	 3 PVC	 4 LDPE	 5 PP	 6 PS
	POLYETHYLENE TEREPHTHALATE	HIGH-DENSITY POLYETHYLENE	POLYVINYL CHLORIDE	LOW-DENSITY POLYETHYLENE	POLYPROPYLENE	POLYSTYRENE
		R=H	Cl	H	CH ₃	fenyl



[1] Callister, W. D., Rethwisch, D. G. (2018). Materials Science and Engineering: An Introduction (10th ed.). Hoboken, NJ: Wiley



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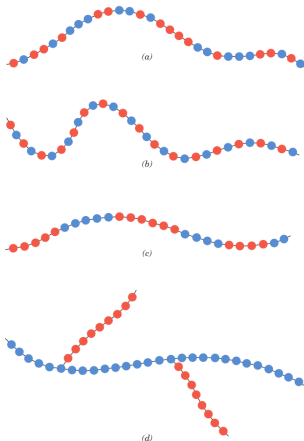
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Type of monomer units

- ▶ Homopolymers
- ▶ Copolymers
 - (a) random
 - (b) alternating
 - (c) block
 - (d) graft

Most popular copolymers:

- ▶ Acrylonitrile Butadiene Styrene (ABS)-Lego
- ▶ Styrene-Butadiene Rubber (SBR)-tires
- ▶ Nylon 6,6-parachutes



[1] Callister, W. D., Rethwisch, D. G. (2018). Materials Science and Engineering: An Introduction (10th ed.). Hoboken, NJ: Wiley



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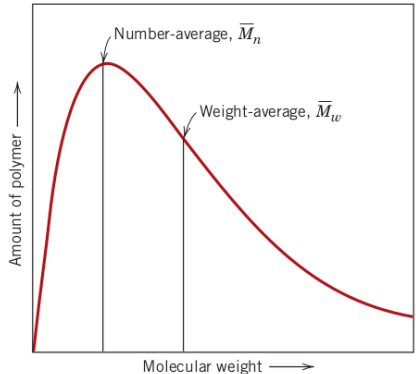
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Molecular weight

- ▶ number-average M_n vs. weight average M_w
- ▶ monodisperse vs. polydisperse samples, $PDI = M_w / M_n$
- ▶ non-entangled vs. entangled
- one of the most important characteristics
 - viscosity $\sim M_w^\alpha$
 - thermal properties M_w dependent



[1] Callister, W. D., Rethwisch, D. G. (2018). Materials Science and Engineering: An Introduction (10th ed.). Hoboken, NJ: Wiley

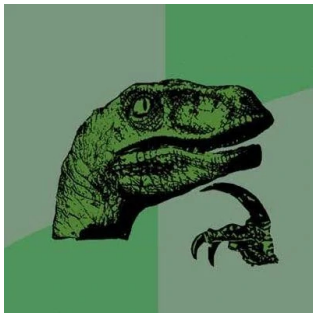


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Guess an average molecular weight of commercial polymers



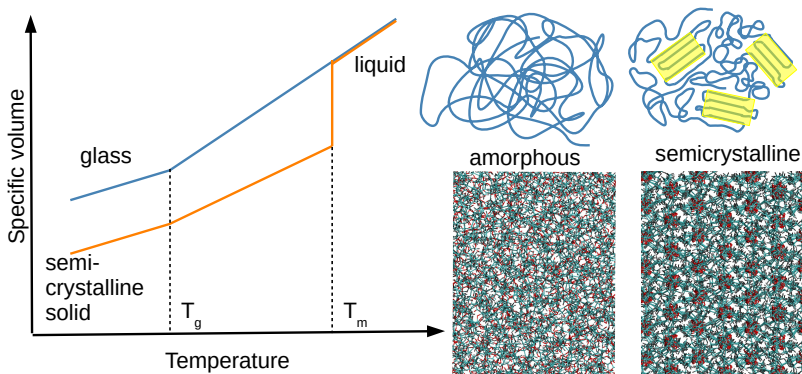
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Amorphous and semicrystalline polymers



→ T_g glass transition temperature

→ T_m melting temperature



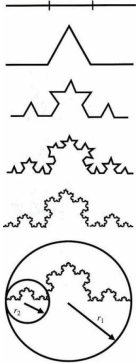
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Fractal character



- ▶ self-similar over a wide range of length scales
- ▶ fractal dimension



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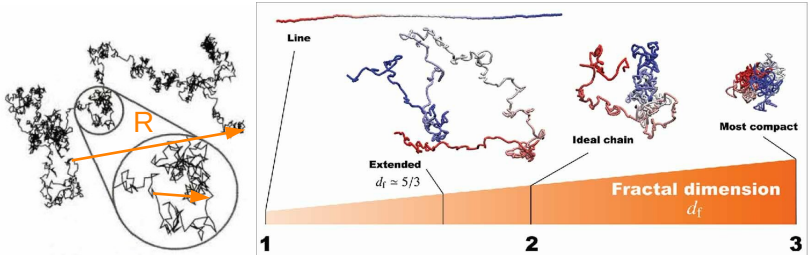
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Fractal character of polymers

- $g \sim (\sqrt{\langle R^2 \rangle})^{d_f}$, g number of monomers in a subsection



!!! configuration highly dependent on environmental factors (temperature, solvent quality, confinement, ionic strength) and intrinsic characteristics (stiffness, internal H-bonding)

[1] Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.

[2] Shinkai, S., Nozaki, T., Maeshima, K., Togashi, Y. (2017). Nucleus, 8(4), 353–359.



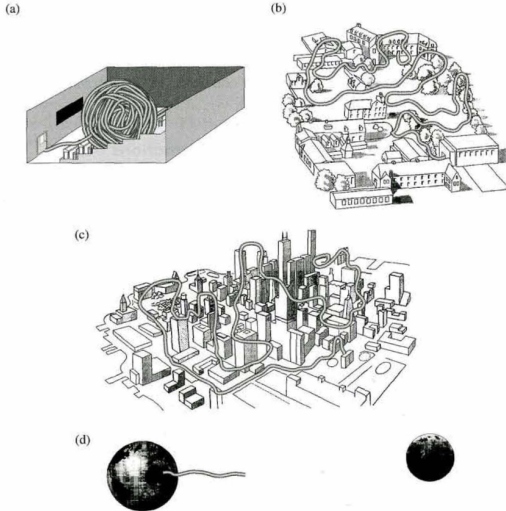
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Fractal character of polymers



$$g = 10^{10}, \text{ size } 1 \text{ cm}$$

- (a) attractive interactions
- (b) zero effective interaction
- (c) short-range repulsion
- (d) long-range repulsion

[1] Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.



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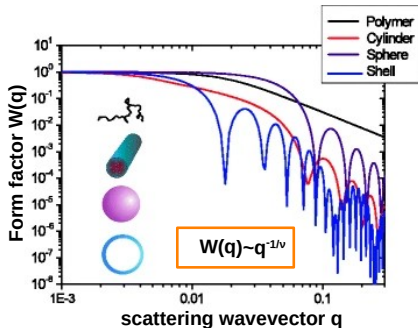
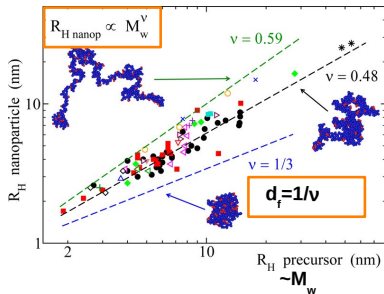
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Fractal character of polymers

- ▶ single-chain form factor $W(q) = \langle \frac{1}{N^2} \sum_{j,k}^N \exp(-i\vec{q}(\vec{R}_j - \vec{R}_k)) \rangle$



[1] Pomposo, J. A.; Perez-Baena, I.; Lo Verso, F.; Moreno, A. J.; Arbe, A.; Colmenero, J.; ACS Macro Letters 2014 3 (8), 767-772

[2] Lund, R., Willner, L., Richter, D. (2013). Controlled Polymerization and Polymeric Structures. Advances in Polymer Science, vol 259. Springer, Cham.



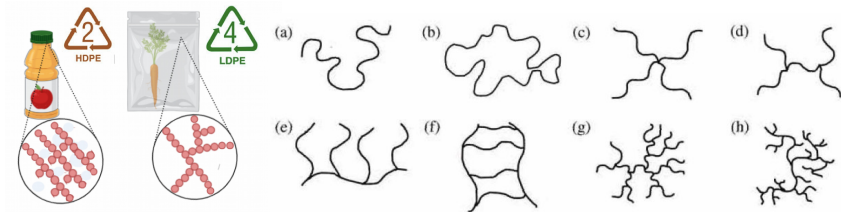
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Polymer architecture



- ▶ **HDPE**: highly crystalline, stiff, hard, high tensile strength
- ▶ **LDPE**: flexible, soft, high toughness and impact strength

- (a) linear
- (b) ring
- (c) star
- (d) H
- (e) comb
- (f) ladder
- (g) dendrimer
- (h) random branching

[*] picture created by BioRender

[1] Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.



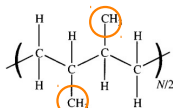
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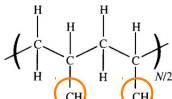


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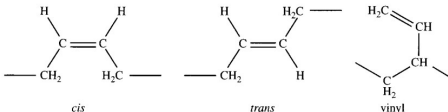
Polymer microstructure



Head-to-head polypropylene

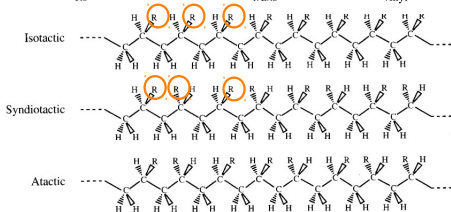


Head-to-tail polypropylene



- ▶ cis, trans and vinyl PB differences in T_m , crystallinity and applications

- ▶ iPP highly crystalline
- ▶ sPP lower crystallinity than iPP
- ▶ aPP amorphous



[1] Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.

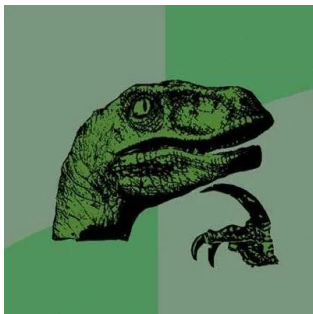


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Find other commercial polymers, which have different stereoisomers.



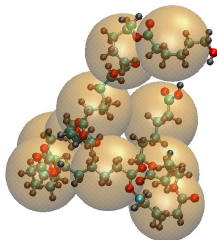
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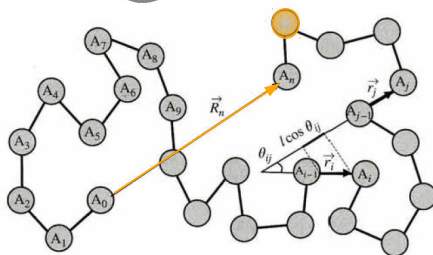
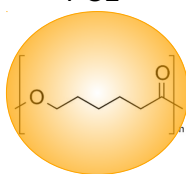


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Theoretical models for polymers



PCL



- ▶ infinite chain with Gaussian statistics
- ▶ no microstructure, coarse-grained picture
- ▶ chemistry-independent
- correct prediction of the large-scale behavior of most real polymers in certain conditions, regardless of their specific chemical structure

[1] Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.



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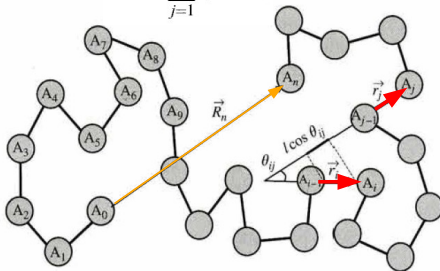
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Conformation of an ideal chain

$$\langle R^2 \rangle = \sum_{i=1}^n \sum_{j=1}^n \langle \vec{r}_i \cdot \vec{r}_j \rangle = l^2 \sum_{i=1}^n \sum_{j=1}^n \langle \cos \theta_{ij} \rangle$$

$$\langle R^2 \rangle = l^2 \sum_{i=1}^n \sum_{j=1}^n \langle \cos \theta_{ij} \rangle = l^2 \sum_{i=1}^n C'_i = C_n n l^2$$

$$C'_i \equiv \sum_{j=1}^n \langle \cos \theta_{ij} \rangle$$



- ▶ fix bond length l
- ▶ correlations in local conformation expressed through cosine
- ▶ if $\langle \cos \theta_{ij} \rangle = 0$ when $i \neq j$ no correlation
- ▶ **end-to-end vector**: if $n = N \Rightarrow R_n = R$
- ▶ **characteristic ratio**: if $\lim_{n \rightarrow \infty} C_n = C_\infty$

[1] Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.

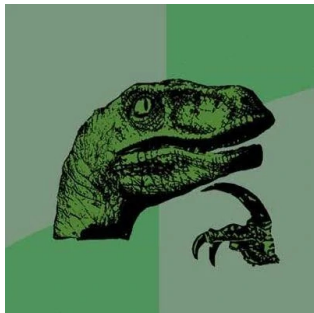


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Why in an ideal isotropic system of multiple chains consisting of n atoms $\langle \vec{R}_n \rangle = 0$?



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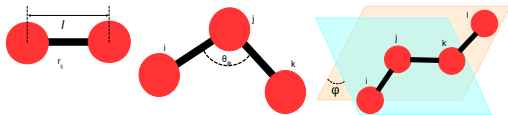


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Ideal models

Flory's characteristic ratio

$$\langle R^2 \rangle = C_{\infty} n l^2$$



Models	FJC	FRC	HR
Bond length l	Fixed	Fixed	Fixed
Bond angle θ	Free	Fixed	Fixed
Torsion angle φ	Free	Free	Controlled by $U(\varphi)$
Next φ independent?	Yes	Yes	Yes
C_{∞}	1	$\frac{1 + \cos \theta}{1 - \cos \theta}$	$\left(\frac{1 + \cos \theta}{1 - \cos \theta} \right) \left(\frac{1 + \langle \cos \theta \rangle}{1 - \langle \cos \varphi \rangle} \right)$

- ▶ **FJC** freely jointed chain
- ▶ **FRC** freely rotating chain
- ▶ **HR** hindered rotation

[1] Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.



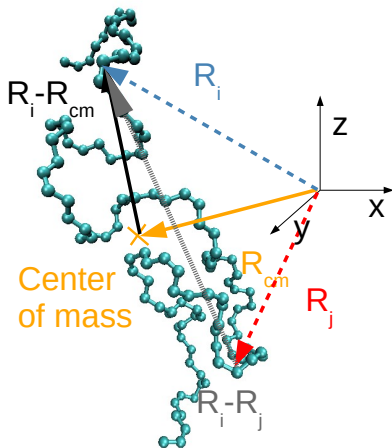
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Radius of gyration of an ideal chain



- definition:

$$R_g^2 = \frac{\sum_i^N m_i (R_i - R_{cm})^2}{\sum_i^N m_i}$$

- if all m_i equal:

$$\langle R_g^2 \rangle = \frac{1}{N} \sum_i^N \langle (R_i - R_{cm})^2 \rangle$$

- $\langle R_g^2 \rangle = \frac{1}{N^2} \sum_i^N \sum_j^N \langle (R_i - R_j)^2 \rangle$

For flexible, Gaussian chains:

$$\langle R_g^2 \rangle = \frac{\langle R^2 \rangle}{6}$$

⇒ indicates ideal behaviour



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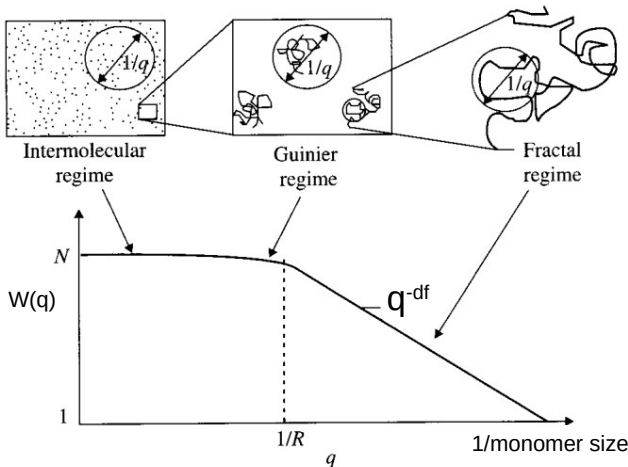
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Scattering Techniques for Polymer Characterization

► $W(q) = \langle \frac{1}{N^2} \sum_{j,k}^N \exp(-i\vec{q}(\vec{R}_j - \vec{R}_k)) \rangle$ $q = \frac{4\pi}{\lambda} \sin(\theta/2)$



Scattering Techniques for Polymer Characterization

- ▶ Small-Angle X-ray Scattering (SAXS): electron density contrast
- ▶ Small-Angle Neutron Scattering (SANS): contrast in scattering length density, often enhanced by deuteration
- ▶ Static Light Scattering (SLS): suitable for large polymers

⇒ estimation of R_g , Guinier function

$$W(q) \cong \exp\left(-\frac{q^2 R_g^2}{3}\right) \quad qR_g < 1$$

+ estimation of aggregation, deformation..

- ▶ Ideal linear chains: form factor expressed by **Debye function**

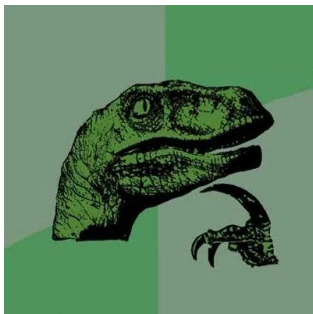


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Suggest a way to collect the scattering from only one molecule



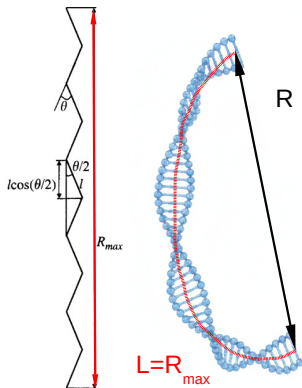
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Worm-like chain model



- ▶ Kratky-Porod model
- ▶ freely rotating chain with small θ angle
- ▶ flexibility related to the fluctuations of the contour length $R_{\max}$
- ▶ semiflexible/stiff polymers
- ▶ persistence length significant in comparison to the total chain length

$$\langle R^2 \rangle = 2l_p \langle R_{\max} \rangle \left(1 - \frac{l_p}{\langle R_{\max} \rangle} [1 - \exp(-\langle R_{\max} \rangle / l_p)] \right)$$

[1] Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.



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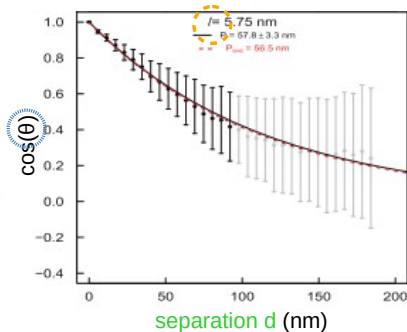
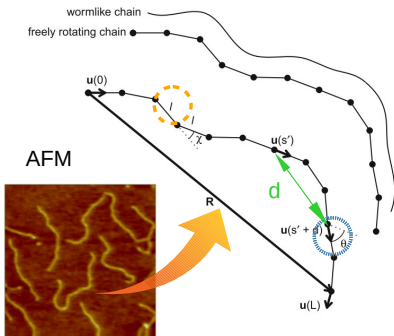
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Kuhn length b , persistence length l_p

$$\langle \cos \theta(d) \rangle = \frac{\vec{u}_i \cdot \vec{u}_{i+d}}{|\vec{u}_i| |\vec{u}_{i+d}|} = \exp(-d/l_p), \quad \text{For WLC} \quad b = 2l_p$$



[1] Peters, J.P., Maher III, L.J. (2024). Approaches for Determining DNA Persistence Length Using Atomic Force Microscopy. In: Dame, R.T. (eds) Bacterial Chromatin. Methods in Molecular Biology



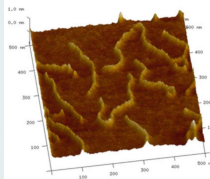
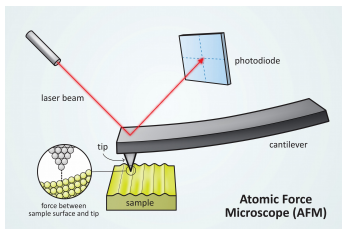
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Atomic Force Microscopy



- ▶ three-dimensional (3D) maps of the polymer surface
- ▶ non-destructive
- ▶ possibility of measuring mechanical properties
- ▶ operability in diverse environments

!! No bulk information

[1] Peters, J.P., Maher III, L.J. (2024). Approaches for Determining DNA Persistence Length Using Atomic Force Microscopy. In: Dame, R.T. (eds) Bacterial Chromatin. Methods in Molecular Biology

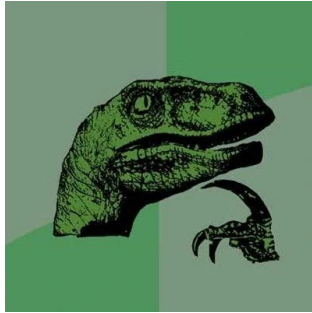


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Your task is to measure the polymer characteristics via AFM. In your opinion, what is the single most critical factor influencing your measurements?



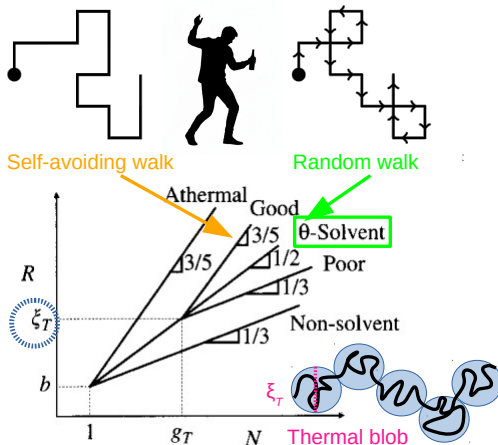
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Real Chains



- ▶ excluded volume
- ▶ thermal blob size ξ_T , EV weaker than kT
- ▶ swollen, ideal and collapsed conformations
- ▶ θ solvent: ideal conformation
- ▶ melt: screened interactions

[1] Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.



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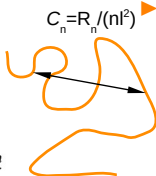
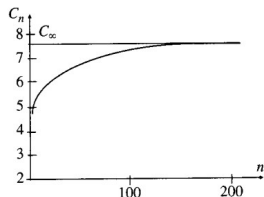


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Chain flexibility: Characteristic ratio

Polymer	C_∞
polypropylene	5.9
atactic polystyrene	9.5
polyethylene	7.4
1,4-polyisoprene	4.6

- ▶ chain-length dependent
- ▶ $\langle R^2 \rangle = C_\infty n l^2$, $C_\infty = \langle R^2 \rangle / n l^2$
- ▶ measurements of R^2 in θ -conditions



- ▶ **Viscometry**: Stockmayer–Fixman equation to extrapolate to ideal conditions, K_Θ proportional to R^2

$$\frac{[\eta]}{M^{1/2}} = K_\Theta + B' M^{1/2}$$

[1] Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.



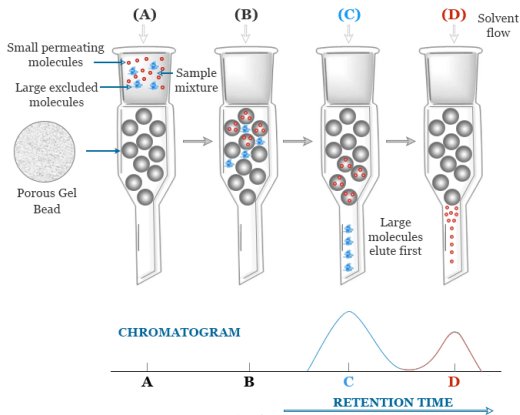
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Gel Permeation Chromatography



- ▶ separation based on hydrodynamic volume
- ▶ elution time correlated with molecular weight
- ▶ calibration to standards or GPC-SLS combination

[1] <https://www.priyamstudycentre.com/2022/02/gel-permeation-chromatography.html>



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Keep calm and stay tuned!

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👉 <https://polysim.eu/>

- ▶ Rubinstein, M.; Colby, R. H., Polymer physics. Oxford university press, 2003.
- ▶ Fetters, L. J., D. J. Lohse, and W. W. Graessley, Chain dimensions and entanglement spacings in dense macromolecular systems, J. Polym. Sci., Polym. Phys. Ed. 37, 1999, 1023–1033
- ▶ Ralf Everaers, Hossein Ali Karimi-Varzaneh, Frank Fleck, Nils Hojdis, and Carsten Svaneborg, Macromolecules, 2020, 53 (6), 1901-1916



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